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MASS TRANSFER IN SOLID-LIQUID EXTRACTION AT HIGH SOLUTE CONCENTRATIONS

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Abstract

The solid-liquid extraction process and some of its influencing factors such as solid-liquid ratio, temperature and salt initial amount have been investigated. Also, mathematical modeling for mass transfer coefficients calculation was applied.

An inert porous solid material (coal), impregnated with 10% and 20% mass NaCl or 15% and 30% mass CaCl₂, was used. The leaching was conducted in a fixed bed column, in laminar flow. The CaCl₂ impregnated samples were investigated in a column with a height/diameter ratio of 2.5, at 20°C, 30°C, and 40°C and the NaCl samples, in a 4.16 height/diameter ratio column, at 30°C, 40°C, 50°C, 60°C. In both cases, liquid flow rates of 3.8 L/h, 7 L/h, 10.6 L/h, 13.3 L/h were used.

An increase of the extraction degree with the washing liquid flow rate (up to 10.6 L/h) indicated that the solid-liquid ratio is a crucial factor. Temperature increase has a positive influence on the extraction degree.

At the beginning of the process, the salt quantity extracted from the high salt amount sample, using the lowest water flow rate has a similar value to the one extracted from the low salt amount sample, using the highest flow rate. At larger time values ($t > 500$ s), the extraction degree dependence on the liquid flow rate is similar for the high and low salt samples.

Based on the proposed mathematical model, mean time mass transfer coefficient values $k \cdot a$ were calculated and compared to the experimental obtained data.

Key words: diffusion, porous materials, kinetics, liquid- solid extraction, mass transfer coefficient

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