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## DECOLORIZING OF PULPING EFFLUENTS BY AMMONIUM LIGNOSULPHONATE OXIDATION

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## **Abstract**

This paper presents the results of studies concerning the oxidative degradation of ammonium lignosulphonate (LS) in the presence of Fenton reagent ( $Fe^{2^+}/H_2O_2$ ) and ammonium persulfate – silver ions systems ( $S_2O_8^{2^-}/Ag^+$ ).

From data obtained, it was found out a more pronounced efficiency of  $\rm S_2O_8^{2-}/Ag^+$  system, discoloration of samples being completed for an LS concentration of 0.63 mg/mL. These results could be also explained by comparing the content in aromatic hydroxylated compounds, which is significantly lower in the case of Fenton reagent, for the same concentration of LS.

The transformations that occur was furthermore evidenced by UV spectroscopy, which allow the quantitative determination of lignin remaining aromatic units based on absorbance reduction at 280 nm, by conductometric characterization of solutions containing oxidized samples, and by turbidity measurements which gives indications concerning the ability of degraded products to precipitate with Ca<sup>2+</sup> ions.

It was also established the optimum conditions like pH, temperature, sample / oxidant ratio, that enable a rapid discoloration of samples.

*Keywords:* decolorization, oxidative degradation, ammonium lignosulphonate, Fenton reagent,  $S_2O_8^{2-}/Ag^+$  system

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