



IMPROVED Fe-BASED PRE-HYDROLYZED COAGULANTS

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Abstract

This paper investigates the conditions in which stable ferric basic solutions, used as coagulants in primary treatments of waters and wastewaters, can be prepared starting from copperas ($\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$). A series of partially hydrolyzed ferric solutions were obtained, using exclusively hydrogen peroxid H_2O_2 as oxidation/basification agent, in a molar ratio $\text{FeSO}_4/\text{H}_2\text{O}_2=1:1$ and different $\text{FeSO}_4/\text{H}_2\text{SO}_4$ molar ratios, from 20:1 to 2:1; temperature was varied from 50°C to 70°C. Simultaneous oxidation and basification lead to partially hydrolyzed Fe(III) solutions, in which the basification degree and stability are controlled by the $\text{FeSO}_4/\text{H}_2\text{SO}_4$ ratio; molar ratios $\text{FeSO}_4/\text{H}_2\text{SO}_4$ between 2.03:1 and 5:1 result in stable basic solutions. Metal-polysilicate coagulants were obtained through a modified copolymerization procedure involving the *in-situ* generation of silicic acid (SA) at a Si/Fe molar ratio of 1:10, followed by oxidation-basification. All the solutions were characterized in terms of total iron concentration, basicity degree, residual Fe(II), density and stability.

Keywords: prehydrolyzed polyferric sulphate, coagulant, metal-polysilicate, turbidity

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